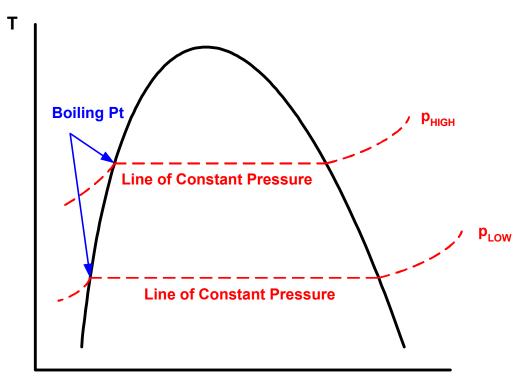
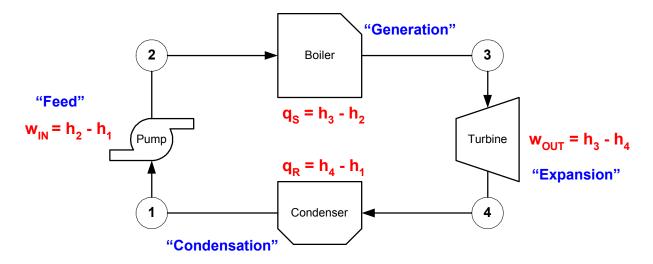


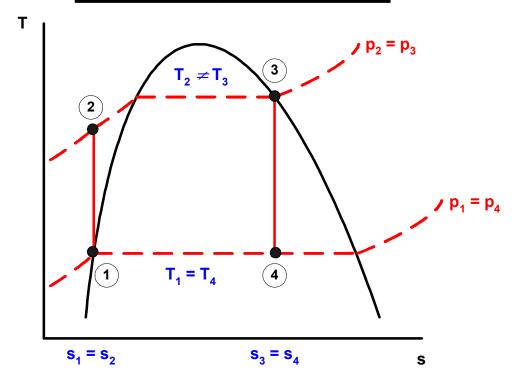
S



Steam Plant Components



Ideal Rankine Cycle (w/o Superheat)



- 1 Saturated Liquid
- 3 Saturated Vapor
- 2 Subcooled Liquid
- 4 Wet Vapor (Saturated Steam)

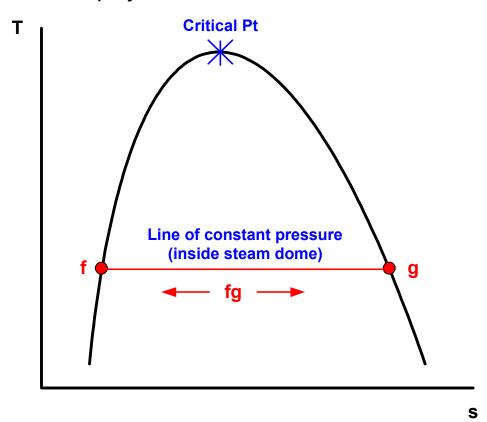
Table 1: Saturated Steam (by Temperature)

	Abs Press.	Specific Volume				Enthalpy					
Temp Fahr t	Lb per Sq In. p	Sat. Liquid V f	Evap V fg	Sat. Vapor Vg	Sat. Liquid h f	Evap h fg	Sat. Vapor h _g	Sat. Liquid ^S f	Evap Sfg	Sat. Vapor S g	Temp Fahr t
32.0*	0.08859	0.016022	3304.7	3304.7	- 0.0179	1075.5	1075.5	0.0000	2.1873	2.1873	32.0 *
34.0	0.09600	0.016021	3061.9	3061.9	1.996	1074.4	1076.4	0.0041	2.1762	2.1802	34.0
36.0	0.10395	0.016020	2839.0	2839.0	4.008	1073.2	1077.2	0.0081	2.1651	2.1732	36.0
38.0	0.11249	0.016019	2634.1	2634.2	6.018	1072.1	1078.1	0.0122	2.1541	2.1663	38.0
40.0	0.12163	0.016019	2445.8	2445.8	8.027	1071.0	1079.0	0.0162	2.1432	2.1594	40.0
42.0	0.13143	0.016019	2272.4	2272.4	10.035	1069.8	1079.9	0.0202	2.1325	2.1527	42.0
44.0	0.14192	0.016019	2112.8	2112.8	12.041	1068.7	1080.7	0.0242	2.1217	2.1459	44.0
46.0	0.15314	0.016020	1965.7	1965.7	14.047	1067.6	1081.6	0.0282	2.1111	2.1393	46.0
48.0	0.16514	0.016021	1830.0	1830.0	16.051	1066.4	1082.5	0.0321	2.1006	2.1327	48.0
50.0	0.17796	0.016023	1704.8	1704.8	18.054	1065.3	1083.4	0.0361	2.0901	2.1262	50.0
52.0	0.19165	0.016024	1589.2	1589.2	20.057	1064.2	1084.2	0.0400	2.0798	2.1197	52.0
54.0	0.20625	0.016026	1482.4	1482.4	22.058	1063.1	1085.1	0.0439	2.0695	2.1134	54.0
56.0	0.22183	0.016028	1383.6	1383.6	24.059	1061.9	1086.0	0.0478	2.0593	2.1070	56.0
58.0	0.23843	0.016031	1292.2	1292.2	26.060	1060.8	1086.9	0.0516	2.0491	2.1008	58.0
60.0	0.25611	0.016033	1207.6	1207.6	28.060	1059.7	1087.7	0.0555	2.0391	2.0946	60.0
62.0	0.27494	0.016036	1129.2	1129.2	30.059	1058.5	1088.6	0.0593	2.0291	2.0885	62.0
64.0	0.29497	0.016039	1056.5	1056.5	32.058	1057.4	1089.5	0.0632	2.0192	2.0824	64.0
66.0	0.31626	0.016043	989.0	989.1	34.056	1056.3	1090.4	0.0670	2.0094	2.0764	66.0
68.0	0.33889	0.016046	926.5	926.5	36.054	1055.2	1091.2	0.0708	1.9996	2.0704	68.0
70.0	0.36292	0.016050	868.3	868.4	38.052	1054.0	1092.1	0.0745	1.9900	2.0645	70.0
72.0	0.38844	0.016054	814.3	814.3	40.049	1052.9	1093.0	0.0783	1.9804	2.0587	72.0
74.0	0.41550	0.016058	764.1	764.1	42.046	1051.8	1093.8	0.0821	1.9708	2.0529	74.0
76.0	0.44420	0.016063	717.4	717.4	44.043	1050.7	1094.7	0.0858	1.9614	2.0472	76.0
78.0	0.47461	0.016067	673.8	673.9	46.040	1049.5	1095.6	0.0895	1.9520	2.0415	78.0
80.0	0.50683	0.016072	633.3	633.3	48.037	1048.4	1096.4	0.0932	1.9426	2.0359	80.0
82.0	0.54093	0.016077	595.5	595.5	50.033	1047.3	1097.3	0.0969	1.9334	2.0303	82.0
84.0	0.57702	0.016082	560.3	560.3	52.029	1046.1	1098.2	0.1006	1.9242	2.0248	84.0
86.0	0.61518	0.016087	527.5	527.5	54.026	1045.0	1099.0	0.1043	1.9151	2.0193	86.0
88.0	0.65551	0.016093	496.8	496.8	56.022	1043.9	1099.9	0.1079	1.9060	2.0139	88.0
90.0 92.0 94.0 96.0	0.69813 0.74313 0.79062 0.84072	0.016099 0.016105 0.016111 0.016117 0.016123	468.1 441.3 416.3 392.8 370.9	468.1 441.3 416.3 392.9 370.9	58.018 60.014 62.010 64.006	1042.7 1041.6 1040.5 1039.3	1100.8 1101.6 1102.5 1103.3	0.1115 0.1152 0.1188 0.1224	1.8970 1.8881 1.8792 1.8704	2.0086 2.0033 1.9980 1.9928	90.0 92.0 94.0 96.0

 Table 2: Saturated Steam (by Pressure)

		Sp	ecific Volu	me		Enthalpy			Entropy		
Abs Press. Lb/Sq In. p	Temp Fahr t	Sat. Liquid V f	Evap V fg	Sat. Vapor v _g	Sat. Liquid h _f	Evap h _{fg}	Sat. Vapor h _g	Sat. Liquid ^S f	Evap S fg	Sat. Vapor s g	Abs Press Lb/Sq In. p
0.08865 0.25 0.50 1.0 5.0 10.0 14.696 15.0	32.018 59.323 79.586 101.74 162.24 193.21 212.00 213.03	0.016022 0.016032 0.016071 0.016136 0.016407 0.016592 0.016719 0.016726	3302.4 1235.5 641.5 333.59 73.515 38.404 26.782 26.274	3302.4 1235.5 641.5 333.60 73.532 38.420 26.799 26.290	0.0003 27.382 47.623 69.73 130.20 161.26 180.17 181.21	1075.5 1060.1 1048.6 1036.1 1000.9 982.1 970.3 969.7	1075.5 1087.4 1096.3 1105.8 1131.1 1143.3 1150.5 1150.9	0.0000 0.0542 0.0925 0.1326 0.2349 0.2836 0.3121 0.3137	2.1872 2.0425 1.9446 1.8455 1.6094 1.5043 1.4447 1.4415	2.1872 2.0967 2.0370 1.9781 1.8443 1.7879 1.7568 1.7552	0.08865 0.25 0.50 1.0 5.0 10.0 14.696 15.0
20.0 30.0 40.0 50.0 60.0 70.0 80.0 90.0	227.96 250.34 267.25 281.02 292.71 302.93 312.04 320.28	0.016834 0.017009 0.017151 0.017274 0.017383 0.017482 0.017573 0.017659	20.070 13.7266 10.4794 8.4967 7.1562 6.1875 5.4536 4.8779	20.087 13.7436 10.4965 8.5140 7.1736 6.2050 5.4711 4.8953	196.27 218.9 236.1 250.2 262.2 272.7 282.1 290.7	960.1 945.2 933.6 923.9 915.4 907.8 900.9 894.6	1156.3 1164.1 1169.8 1174.1 1177.6 1180.6 1183.1 1185.3	0.3358 0.3682 0.3921 0.4112 0.4273 0.4411 0.4534 0.4643	1.3962 1.3313 1.2844 1.2474 1.2167 1.1905 1.1675 1.1470	1.7320 1.6995 1.6765 1.6586 1.6440 1.6316 1.6208 1.6113	20.0 30.0 40.0 50.0 60.0 70.0 80.0
100.0 110.0 120.0 130.0 140.0 150.0 160.0 170.0 180.0	327.82 334.79 341.27 347.33 353.04 358.43 363.55 368.42 373.08 377.53	0.017740 0.01782 0.01789 0.01796 0.01803 0.01809 0.01815 0.01821 0.01827 0.01833	4.4133 4.0306 3.7097 3.4364 3.2010 2.9958 2.8155 2.6556 2.5129 2.3847	4.4310 4.0484 3.7275 3.4544 3.2190 3.0139 2.8336 2.6738 2.5312 2.4030	298.5 305.8 312.6 319.0 325.0 330.6 336.1 341.2 346.2 350.9	888.6 883.1 877.8 872.8 868.0 863.4 859.0 854.8 850.7 846.7	1187.2 1188.9 1190.4 1191.7 1193.0 1194.1 1195.1 1196.0 1196.9 1197.6	0.4743 0.4834 0.4919 0.4998 0.5071 0.5141 0.5206 0.5269 0.5328 0.5384	1.1284 1.1115 1.0960 1.0815 1.0681 1.0554 1.0435 1.0322 1.0215 1.0113	1.6027 1.5950 1.5879 1.5813 1.5752 1.5695 1.5641 1.5591 1.5543 1.5498	100.0 110.0 120.0 130.0 140.0 150.0 160.0 170.0 180.0
200.0 210.0 220.0 230.0 240.0 250.0 270.0 280.0 290.0	381.80 385.91 389.88 393.70 397.39 400.97 404.44 407.80 411.07 414.25	0.01839 0.01844 0.01850 0.01855 0.01860 0.01865 0.01870 0.01875 0.01880 0.01885	2.2689 2.16373 2.06779 1.97991 1.89909 1.82452 1.75548 1.69137 1.63169 1.57597	2.2873 2.18217 2.08629 1.99846 1.91769 1.84317 1.77418 1.71013 1.65049 1.59482	355.5 359.9 364.2 368.3 372.3 376.1 379.9 383.6 387.1 390.6	842.8 839.1 835.4 831.8 828.4 825.0 821.6 8118.3 815.1 812.0	1198.3 1199.0 1199.6 1200.1 1200.6 1201.1 1201.5 1201.9 1202.3 1202.6	0.5438 0.5490 0.5540 0.5588 0.5634 0.5679 0.5722 0.5764 0.5805	1.0016 0.9923 0.9834 0.9748 0.9665 0.9585 0.9508 0.9433 0.9361 0.9291	1.5454 1.5413 1.5374 1.5336 1.5299 1.5264 1.5230 1.5197 1.5166 1.5135	200.0 210.0 220.0 230.0 240.0 250.0 260.0 270.0 280.0

Property Values On or Inside the Steam Dome



Looking Up f and g Values from a T-s Diagram in the Steam Tables 1 and 2

f is any point on the steam dome left of the critical point - saturated liquid

g is any point on the steam dome right of the critical point - saturated vapor

f and g are both at the same pressure (and notice, both at the same temperature)

fg is the difference ("distance") between value g and value f, for example:

$$s_g - s_f = s_{fg}$$

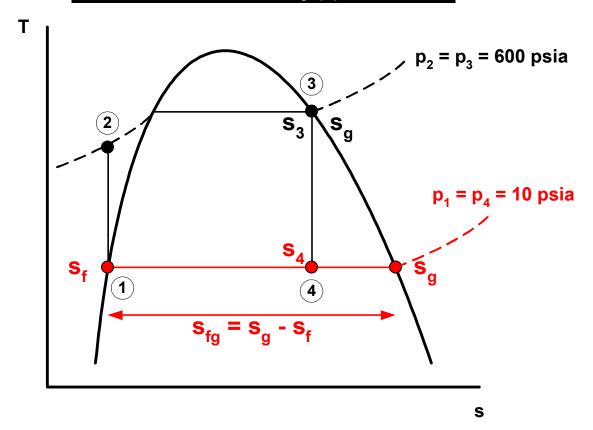
 $h_g - h_f = h_{fg}$

$$n_g$$
 - n_f = n_{fg}

all values between f and g are in the saturated steam (wet vapor) phase

Steam Tables 1 and 2 are the only property tables you need to solve steam cycle problems that do not have superheat.

Steam Quality (x)



"Steam Quality" is denoted by the variable x. It represents the percentage of vapor (0% < x < 100%) in a WET VAPOR state point.

In the above T-s diagram, state point 4 is a wet vapor. This state point has a steam quality (x) given by:

$$s_f + (x_4)s_{fg} = s_4$$

Note that $3 \rightarrow 4$ is an isentropic expansion in the turbine (i.e. $s_3 = s_4$). So, s_3 is easily looked up as the s_g value for 600 psia in the steam tables.

Then we can look up s_f and s_{fg} values for 10 psia. Then solve for x:

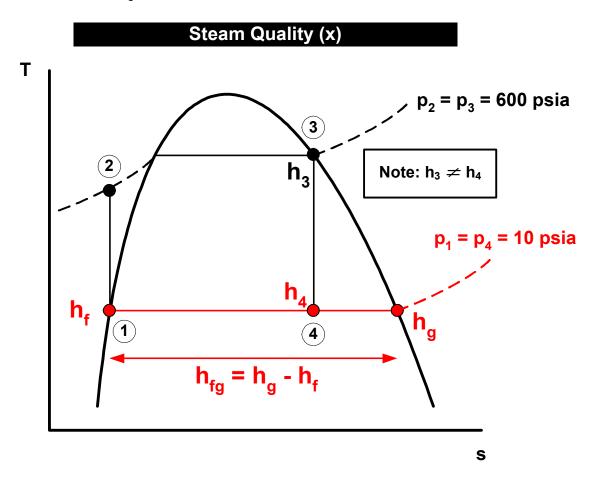
$$x_4 = \frac{S_4 - S_f}{S_{fg}}$$
 %, where $S_4 = S_3$

The counterpart variable to x is "Moisture Content" denoted by the variable m, and given by m = 100% - x. It indicates the <u>percentage of liquid</u> (0% < m < 100%) in a WET VAPOR state point. (Note: Use of m is required for the <u>Mollier Diagram</u>, which is an h-s diagram.)

The reason we solve for x is to find the value of h₄ using this same technique:

$$h_f + (x)h_{fg} = h_4$$

The above h_{f} and h_{fg} values for 10 psia are looked up in the steam tables.



Thus far we know how to solve for h_1 , h_3 , and h_4 .

How about solving for h_2 ?

Process 1 \rightarrow 2 is work in (pump work). This is given by: $w_{12} = h_1 - h_2$ or $w_{PUMP} = |w_{12}| = h_2 - h_1$

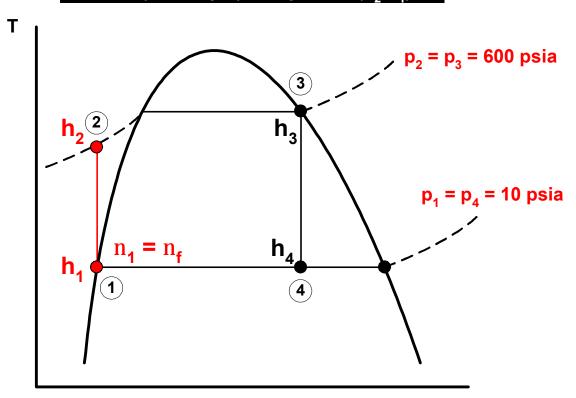
In the ideal Rankine cycle, it is an <u>isentropic</u> process, which makes it an ideal pump (i.e. 100% component efficiency). However, we <u>will not</u> be setting $s_1 = s_2$ to solve for h_2 , due to the complexities of using the subcooled liquid tables (which you do not have).

Remember that during $1 \rightarrow 2$, the water is in a pure liquid phase (saturated liquid or subcooled liquid) – that is, it is incompressible. "Incompressible" means that the water has a constant specific volume, ν (assume $\nu_1 = \nu_2 = \nu_f$ @ p₁). From a SFEE balance for an isentropic process, it is found that:

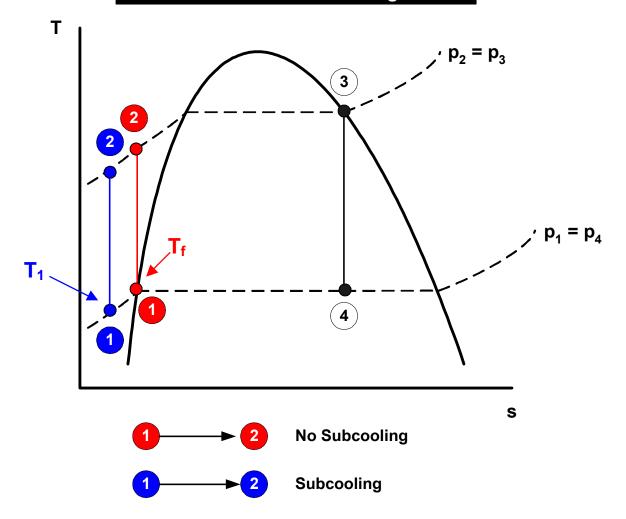
$$w_{12} = \nu_1 (p_1 - p_2) = h_1 - h_2 \underline{or} w_{PUMP} = |w_{12}| = \nu_1 (p_2 - p_1) = h_2 - h_1$$

Thus, the key to solving for h_2 is identifying ν_1 (simply look up $\nu_1 = \nu_f @ p_1$) and identifying p_1 and p_2 . (Note there is an exception in the case of *condenser* subcooling where ν_1 is looked up at $T_1 = T_f - T_{SUBCOOL}$ instead.)

Ideal (isentropic) Pump Work (h₂-h₄)



Condenser Subcooling



In the case of *condenser subcooling*, the water leaves the condenser at a temperature lower than the saturated liquid temperature (T_f) of the ideal Rankine cycle. HOWEVER, since the condenser is modeled as isobaric (i.e. $p_1 = p_4$), the new state point 1 is <u>still</u> along the constant pressure line, just in the subcooled liquid region at a LOWER temperature than T_f .

Effect of subcooling:

If condenser subcooling occurs (usually given as something like "the condensate leaves the condenser with 7°F subcooling"), then ν_1 (and h_1) MUST NOT be

looked up at pressure p_1 in the steam table. Instead ν_1 (and h_1) must be looked up at the temperature $T_1 = T_f$ (@p1) – $T_{SUBCOOL}$. Since you do not have subcooled liquid tables, you will approximate by using the value ν_f (@ T_1) = ν_1 .

Summary of enthalpy values (<u>ideal</u> Rankine cycle):
h ₁ = h _f (@ p ₁ = p ₄ pressure in steam tables), except subcooling
✓ $h2 = h1 + \nu_1 (p_2 - p_1) = h1 + w_{PUMP}$
$$ $h_3 = h_g$ (@ $p_2 = p_3$ pressure in steam tables)
h ₄ = h _f + (x) h _{fg} (h _f and h _{fg} are @ p ₁ = p ₄ pressure in steam tables)
● [™] NOTES:

(1) DO NOT USE the pump technique for the turbine enthalpies. That is:

$$h_3 \neq h_4 + \nu_4 (p_3 - p_4)$$

because, in the turbine, the water IS NOT an incompressible liquid!

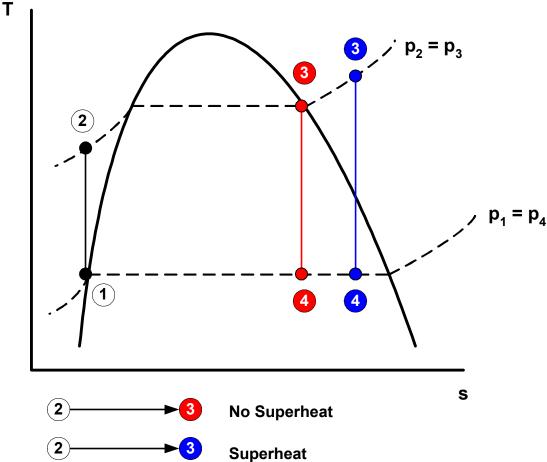
(2) Remember pump work in the form " ν_1 (p₂ – p₁)" does not directly work out to units of [Btu/Ib_m], so use appropriate conversion factors.

(3) $\nu_1 = \nu_f$ will NOT be looked up at $p_1 = p_4$ IF there is *condenser subcooling*. Use ν_1 @ T_1 instead, where $T_1 = T_f$ (@p1) – $T_{SUBCOOL}$.

FOR ANALYSIS OF A STEAM CYCLE, a state point properties table will be useful. For the <u>ideal</u> Rankine cycle, we have 4 state points, thus:

	1	2	3	4
p [psia]				
T [°F]				
h [Btu/lbm]				
s [Btu/lb _m °R]				
$ u$ [ft 3 /lb $_{ m m}$]				





Superheat is the result of the boiler supplying enough heat $(q_s = h_3 - h_2)$ such that state point 3 (exit of the boiler) exceeds the saturation temperature $(T_f = T_g)$ of the water at boiler pressure $(p_2 = p_3)$. In other words, the steam leaves the boiler in a superheated vapor phase.

In finding h_3 , the only difference now is that we must use the superheated steam portion (table 3) of the steam tables. Without superheat, previously, we would use the saturated steam (table 1 or table 2) portion of the steam tables.

Table 3: Superheated Steam (by Pressure)

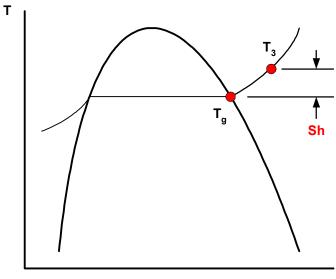
Table 3. Superheated Steam

Abs Press. Lb/Sq In.		0-4	Sat Temperature — Degrees Fahrenheit														
(Sat. Temp)		Sat. Water	Sat. Steam	200	250	300	350	400	450	500	600	700	800	900	1000	1100	1200
1 (101.74)	Sh v h s	0.01614 69.73 0.1326	333.6 1105.8 1.9781	98.26 392.5 1150.2 2.0509	148.26 422.4 1172.9 2.0841	198.26 452.3 1195.7 2.1152	248.26 482.1 1218.7 2.1445	298.26 511.9 1241.8 2.1722	348.26 541.7 1265.1 2.1985	398.26 571.5 1288.6 2.2237	498.26 631.1 1336.1 2.2708	598.26 690.7 1384.5 2.3144	698.26 750.3 1433.7 2.3551	798.26 809.8 1483.8 2.3934	898.26 869.4 1534.9 2.4296	998.26 929.0 1586.8 2.4640	1098.26 988.6 1639.7 2.4969
5 (162.24)	Sh v h s	0.01641 130.20 0.2349	73.53 1131.1 1.8443	37.76 78.14 1148.6 1.8716	87.76 84.21 1171.7 1.9054	137.76 90.24 1194.8 1.9369	187.76 96.25 1218.0 1.9664	237.76 102.24 1241.3 1.9943	287.76 108.23 1264.7 2.0208	337.76 114.21 1288.2 2.0460	437.76 126.15 1335.9 2.0932	537.76 138.08 1384.3 2.1369	637.76 150.01 1433.6 2.1776	737.76 161.94 1483.7 2.2159	837.76 173.86 1534.7 2.2521	937.76 185.78 1586.7 2.2866	1037.76 197.70 1639.6 2.3194
10 (193.21)	Sh v h s	0.01659 161.26 0.2836	38.42 1143.3 1.7879	6.79 38.84 1146.6 1.7928	56.79 41.93 1170.2 1.8273	106.79 44.98 1193.7 1.8593	156.79 48.02 1217.1 1.8892	206.79 51.03 1240.6 1.9173	256.79 54.04 1264.1 1.9439	306.79 57.04 1287.8 1.9692	406.79 63.03 1335.5 2.0166	506.79 69.00 1384.0 2.0603	606.79 74.98 1433.4 2.1011	706.79 80.94 1483.5 2.1394	806.79 86.91 1534.6 2.1757	906.79 92.87 1586.6 2.2101	1006.79 98.84 1639.5 2.2430
1 4.696 (212.00)	Sh v h s	.0167 180.17 .3121	26.799 1150.5 1.7568		38.00 28.42 1168.8 1.7833	88.00 30.52 1192.6 1.8158	138.00 32.60 1216.3 1.8459	188.00 34.67 1239.9 1.8743	238.00 36.72 1263.6 1.9010	288.00 38.77 1287.4 1.9265	388.00 42.86 1335.2 1.9739	488.00 46.93 1383.8 2.0177	588.00 51.00 1433.2 2.0585	688.00 55.06 1483.4 2.0969	788.00 59.13 1534.5 2.1332	888.00 63.19 1586.5 2.1676	988.00 67.25 1639.4 2.2005
15 (213.03)	Sh v h s,	0.01673 181.21 0.3137	26.290 1150.9 1.7552		36.97 27.837 1168.7 1.7809	86.97 29.899 1192.5 1.8134	136.97 31.939 1216.2 1.8437	186.97 33.963 1239.9 1.8720	236.97 35.977 1263.6 1.8988	286.97 37.985 1287.3 1.9242	386.97 41.986 1335.2 1.9717	486.97 45.978 1383.8 2.0155	586.97 49.964 1433.2 2.0563	686.97 53.946 1483.4 2.0946	786.97 57.926 1534.5 2.1309	886.97 61.905 1586.5 2.1653	986.97 65.882 1639.4 2.1982
20 (227.96)	Sh v h s	0.01683 196.27 0.3358	20.087 1156.3 1.7320		22.04 20.788 1167.1 1.7475	72.04 22.356 1191.4 1.7805	122.04 23.900 1215.4 1.8111	172.04 25.428 1239.2 1.8397	222.04 26.946 1263.0 1.8666	272.04 28.457 1286.9 1.8921	372.04 31.466 1334.9 1.9397	472.04 34.465 1383.5 1.9836	572.04 37.458 1432.9 2.0244	672.04 40.447 1483.2 2.0628	772.04 43.435 1534.3 2.0991	872.04 46.420 1586.3 2.1336	972.04 49.405 1639.3 2.1665
25 (240.07)	Sh v h s	0.01693 208.52 0.3535	16.301 1160.6 1.7141		9.93 16.558 1165.6 1.7212	59.93 17.829 1190.2 1.7547	109.93 19.076 1214.5 1.7856	159.93 20.307 1238.5 1.8145	209.93 21.527 1262.5 1.8415	259.93 22.740 1286.4 1.8672	359.93 25.153 1334.6 1.9149	459.93 27.557 1383.3 1.9588	559.93 29.954 1432.7 1.9997	659.93 32.348 1483.0 2.0381	759.93 34.740 1534.2 2.0744	859.93 37.130 1586.2 2.1089	959.93 39.518 1639.2 2.1418
30 (250.34)	Sh v h s	0.01701 218.93 0.3682	13.744 1164.1 1.6995			49.66 14.810 1189.0 1.7334	99.66 15.859 1213.6 1.7647	149.66 16.892 1237.8 1.7937	199.66 17.914 1261.9 1.8210	249.66 18.929 1286.0 1.8467	349.66 20.945 1334.2 1.8946	449.66 22.951 1383.0 1.9386	549.66 24.952 1432.5 1.9795	649.66 26.949 1482.8 2.0179	749.66 28.943 1534.0 2.0543	849.66 30.936 1586.1 2.0888	949.66 32.927 1639.0 2.1217

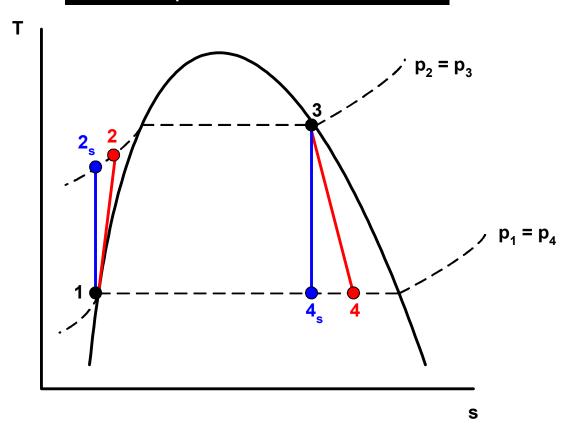
Table 3 has a different layout than Tables 1 and 2.

Generally, you will look up property values by using the <u>boiler pressure</u> as your entering argument. Then use the <u>maximum boiler temperature</u> as your cross-reference argument to find n, h, or s.

What is Sh? This is the "degrees of superheat." In other words, this is the number of degrees (Fahrenheit) above the saturation temperature T_g .



Pump and Turbine Efficiencies



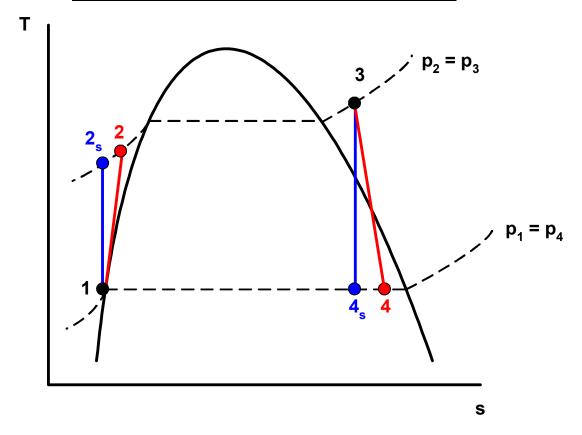
Component efficiencies for the <u>pump</u> and <u>turbine</u> are similar to that of the compressor and turbine in the gas turbine engine:

$$h_{PUMP} = \frac{h_{2s} - h_{1}}{h_{2} - h_{1}}$$
 and $h_{TURB} = \frac{h_{3} - h_{4}}{h_{3} - h_{4s}}$

The component efficiency of the <u>boiler</u> is very similar to the component efficiency of the combustion chamber in the GT engine:

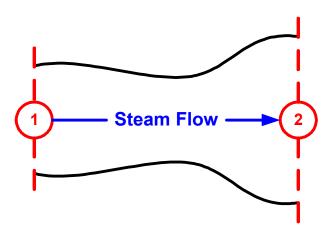
$$h_{\text{BOILER}} = \frac{\dot{m}_{\text{STM}}(\textit{h}_{3} - \textit{h}_{2})}{\dot{m}_{\text{FUEL}}(\textit{HHV})}$$

Pump and Turbine Efficiencies (w/ superheat)



Note: Look at h_4 and h_{4s} . Inside the steam dome both points lie on the constant pressure line $p_1 = p_4$. It may not be apparent inside the steam dome, but we find that $(h_3 - h_4) < (h_3 - h_{4s})$ because $h_{4s} < h_4$.

Nozzle



Nozzles are used inside the turbine to produce a high velocity steam that strikes the turbine blades (which in turns causes turbine rotation to produce shaft output power).

Remember that the working fluid, as it enters the turbine (i.e. exits the boiler), is now a <u>compressible</u> saturated vapor or superheated vapor. If a nozzle is designed properly, it is possible to increase the <u>velocity V</u> (of a compressible fluid) as the nozzle <u>cross-section area A increases</u>. So we find for the turbine nozzle:

$$A_2 > A_1$$
 and $V_2 > V_1$

It is useful in turbine design to analyze the SFEE properties of the turbine nozzle:

$$\left(\frac{g}{g_c}\right) z_1 + \left(\frac{1}{2g_c}\right) V_1^2 + h_1 + q_{12} = \left(\frac{g}{g_c}\right) z_2 + \left(\frac{1}{2g_c}\right) V_2^2 + h_2 + w_{12}$$

By assuming an adiabatic ($q_{12} = 0$) nozzle design with negligible height difference ($z_1 = z_2$) and no mechanical work done ($w_{12} = 0$), we are left with the following:

$$V_2^2 - V_1^2 = 2g_c(h_1 - h_2)$$

NOTES

- (1) DO NOT confuse the above h_1 and h_2 for the enthalpies at the entrance and exit of the pump. These are just the enthalpies at the entrance (1) and exit (2) of the nozzle.
- (2) Usually, but not always, velocity V_1 is negligible compared to velocity V_2 . You must carefully read the problem statement to determine if you can neglect V_1 .